

Gas emissions in thermal propagation experiments

26th GTR EVS meeting 18th-20th of April 2023

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Devices under test (DUTs) – Vehicle and Pack





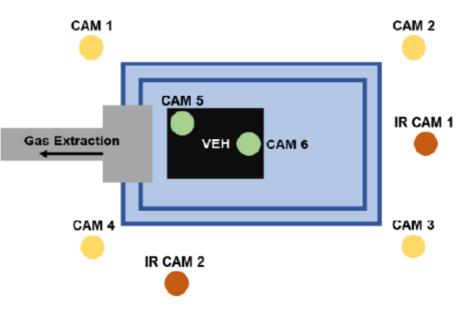
- Factory-new contemporary commerciallyavailable electric vehicle and 2 packs
- Li-ion battery
- 52 kWh, 12 modules, 192 cells
- Pouch cells
- NMC 622 / graphite chemistry
- MY 2021
- Tested in under conditions closely matching "(c) temporary parking" mode

EVS24-E1TP-0300 [EC]JRC's thermal runaway propagation test campaign at .pdf



Experimental Set-up





Stainless steel pool.

Aluminium extraction hood, with glass fibre filter.

Smoke gas ventilator.





Experimental Set-up



Washing bottles and stainless steel canisters.



Open Path Gas Imaging Spectrometer



Online Monitoring FTIR





Time sequence

Pack



Event	Run time (sec)	
	Раск #1	VEHICLE
Start TRIM	0	0
Gas – 1 st venting	23	22
Time to smoke	23	22
Gas – 2 nd venting	74	97
Gas – 3 rd venting	129	164
Fire breaks out	392	534
Fire quenching	1625	1355

Vehicle





Pack



7 seconds after 1st vent

Event	Run tin	Run time (sec)	
	Раск #1	VEHICLE	
Start TRIM	0	0	
Gas – 1 st venting	23	22	
Time to smoke	23	22	
Gas – 2 nd venting	74	97	
Gas – 3 rd venting	129	164	
Fire breaks out	392	534	
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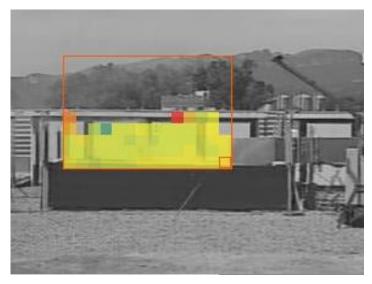
Vehicle



9 seconds after 1st vent



Pack

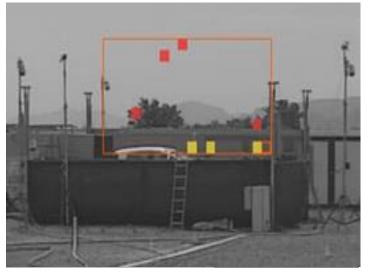


Compounds detected: Diethyl Carbonate Dimethyl Carbonate Ethylmethyl Carbonat Ethylene **Compounds suspected:** Ethylene Carbonate; Methanol

Event	Run time (sec)	
	Раск #1	VEHICLE
Start TRIM	0	0
Gas – 1 st venting	23	22
Time to smoke	23	22
Gas – 2 nd venting	74	97
Gas – 3 rd venting	129	164
Fire breaks out	392	534
Fire quenching	1625	1355



Vehicle



Compounds suspected: Ethylene Carbonate Dimethyl Carbonate



Pack



13 seconds after 2nd vent;

Event	Run tin	ne (sec)
	Раск #1	VEHICLE
Start TRIM	0	0
Gas – 1 st venting	23	22
Time to smoke	23	22
Gas – 2 nd venting	74	97
Gas – 3 rd venting	129	164
Fire breaks out	392	534
Fire quenching	1625	1355

Vehicle



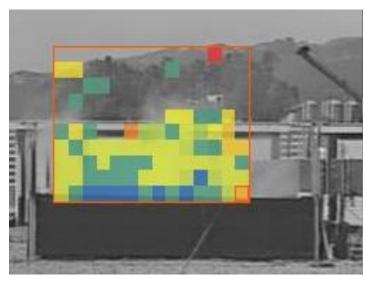
5 seconds after 2nd vent;



Event

Fire quenching

Pack



Compounds detected: CompoundsDiethyl Carbonatesuspected:Dimethyl CarbonateEthylene Carbonate;Ethylmethyl CarbonateMethanolEthyleneState State Stat

PACK #1 VEHICLE Start TRIM 0 0 23 22 Gas – 1st venting Time to smoke 23 22 Gas – 2nd venting 74 97 Gas – 3rd venting 129 164 Fire breaks out 392 534

Run time (sec)

1625

1355

Vehicle



Compounds detected: CompoundsDiethyl Carbonatesuspected:Dimethyl CarbonateEthylene Carbonate;Ethylmethyl CarbonateMethanolEthyleneState State Stat



Pack



6 seconds after 3rd vent

Event	Run time (sec)	
	Раск #1	VEHICLE
Start TRIM	0	0
Gas – 1 st venting	23	22
Time to smoke	23	22
Gas – 2 nd venting	74	97
Gas – 3 rd venting	129	164
Fire breaks out	392	534
Fire quenching	1625	1355

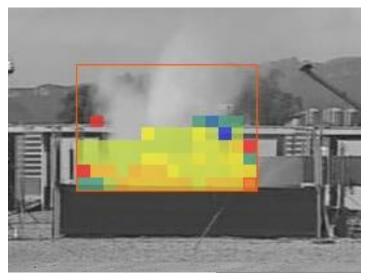
Vehicle



7 seconds after 3rd vent



Pack

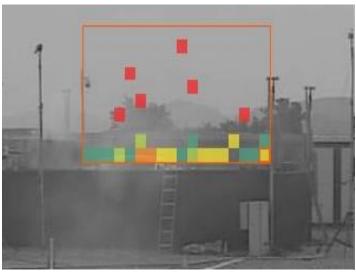


Compounds detected: CompoundsDiethyl Carbonatesuspected:Dimethyl CarbonateEthylene CarboEthylmethyl CarbonateEthyleneMethanolState State Sta

Run time (sec) **Event PACK #1** VEHICLE Start TRIM 0 0 Gas – 1st venting 23 22 Time to smoke 23 22 Gas – 2nd venting 74 97 Gas – 3rd venting 129 164 Fire breaks out 392 534 Fire quenching 1625 1355



Vehicle



Compounds detected: Diethyl Carbonate Dimethyl Carbonate Ethylmethyl Carbonate Ethylene

Compounds suspected: Ethylene Carbonate Methanol



Pack



13 seconds after fire breaking

Event	Run tin	Run time (sec)	
	Раск #1	VEHICLE	
Start TRIM	0	0	
Gas – 1 st venting	23	22	
Time to smoke	23	22	
Gas – 2 nd venting	74	97	
Gas – 3 rd venting	129	164	
Fire breaks out	392	534	
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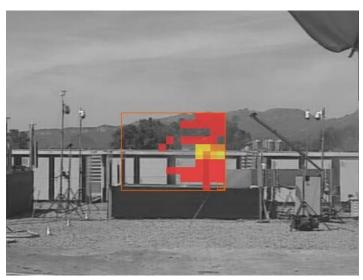
Vehicle

7 seconds after fire breaking

Water screen used by firefighters is visible.



Pack



Compounds detected: Diethyl Carbonate Ethylmethyl Carbonate Methanol Ethylene Carbonate

Event	Run time (sec)	
	Раск #1	VEHICLE
Start TRIM	0	0
Gas – 1 st venting	23	22
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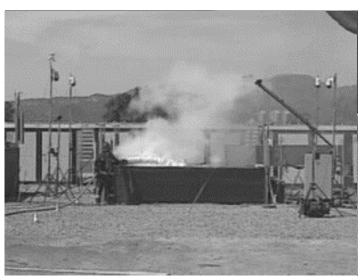




Compounds suspected: Ethylene Carbonate



Pack



10 seconds into fire fighting. Fire brigade uses foam to fill the test pool.

Event	Run tin	ne (sec)
	Раск #1	VEHICLE
Start TRIM	0	0
Gas – 1 st venting	23	22
Time to smoke	23	22
Gas – 2 nd venting	74	97
Gas – 3 rd venting	129	164
Fire breaks out	392	534
Fire quenching	1625	1355

Vehicle



15 seconds into fire fighting. Fire brigade uses water to fill the test pool.



Pack



Compounds detected: Dimethyl Carbonate Ethylmethyl Carbonate Ethylene Carbonate

Event	Run time (sec)	
	Раск #1	VEHICLE
Start TRIM	0	0
Gas – 1 st venting	23	22
Time to smoke	23	22
Gas – 2 nd venting	74	97
Gas – 3 rd venting	129	164
Fire breaks out	392	534
Fire quenching	1625	1355



Vehicle



Compounds suspected: Dimethyl Carbonate



Open path FTIR so far...

IDENTIFIED	SUSPECTED
Dimethyl Carbonate	Ethylene Carbonate
Diethyl Carbonate	Methanol
Ethylmethyl Carbonate	
Ethylene	
Methanol	
Ethylene Carbonate	

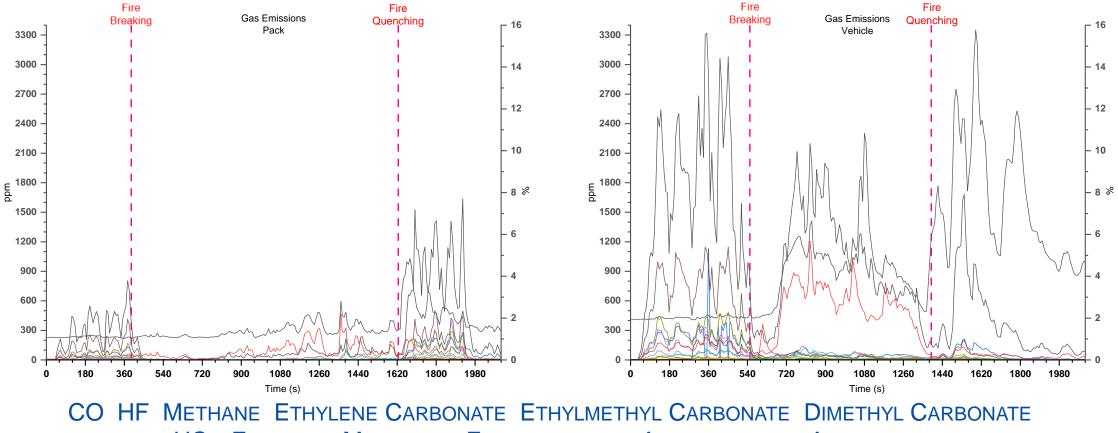
- Allows following evolution of gases in the smoke cloud...
- ...From a (safe) distance!



IDENTIFIED	SUSPECTED
Dimethyl Carbonate	Ethylene Carbonate
Diethyl Carbonate	Methanol
Ethylmethyl Carbonate	Dimethyl Carbonate
Ethylene	



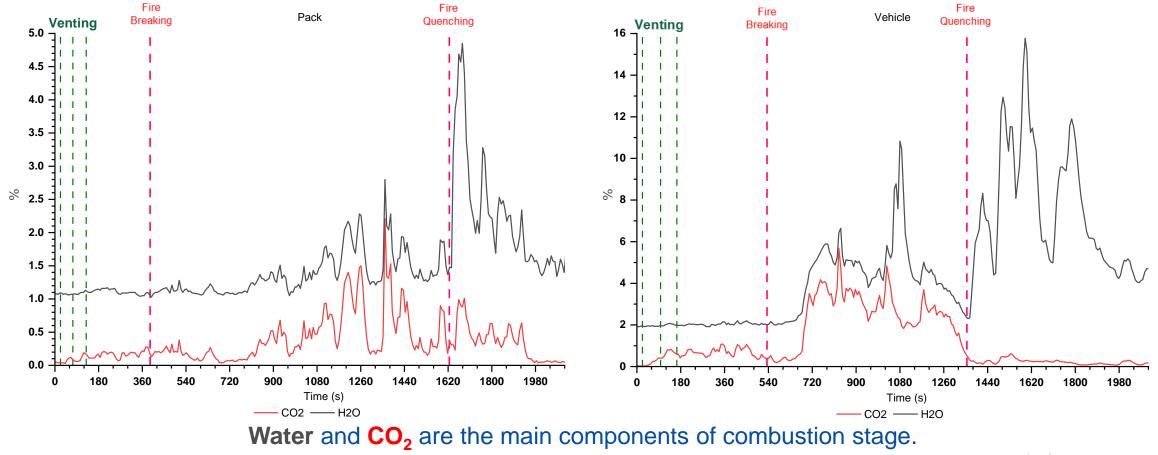




HCL ETHYLENE METHANOL FORMALDEHYDE ACETALDEHYDE ACETYLENE

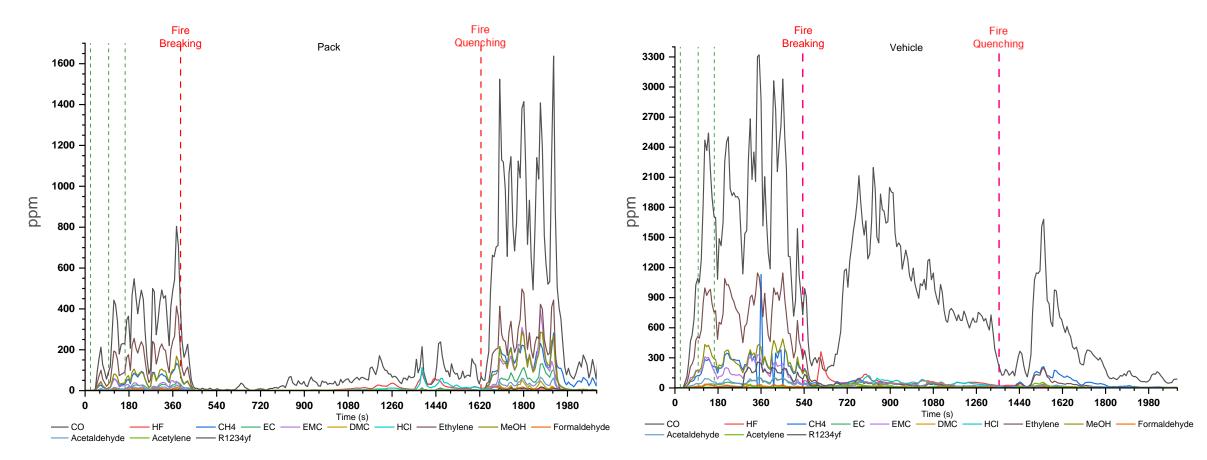
R1234YF WATER CO2





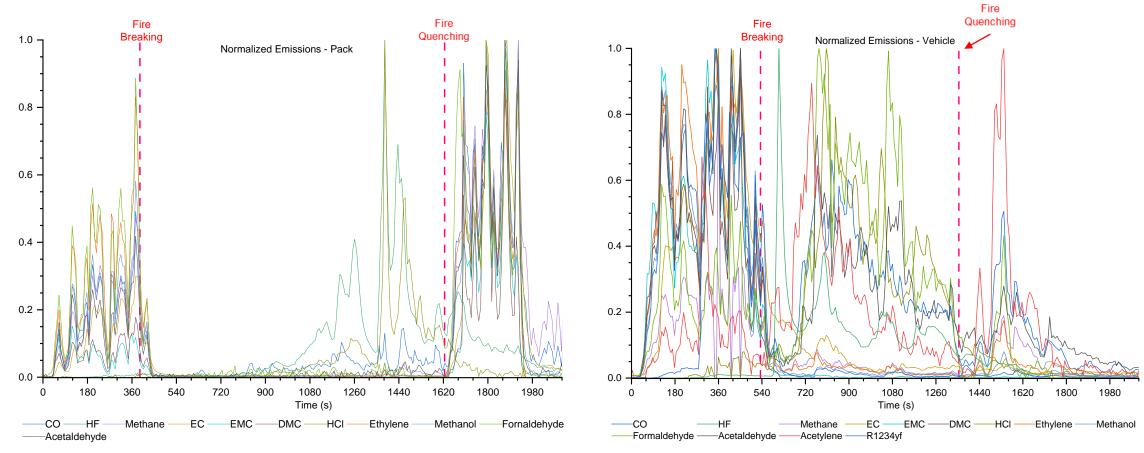
They show different behaviour in fire quenching stage.





Up to 13 species. Similar pre combustion profiles at different scale?

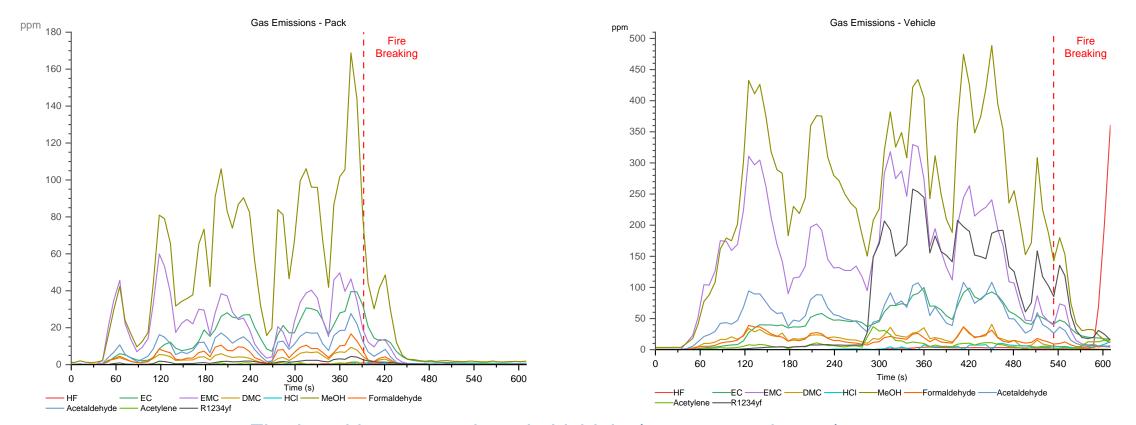




Different pattern of emissions. Not only matter of scale!



Gas Analysis – First 10 minutes

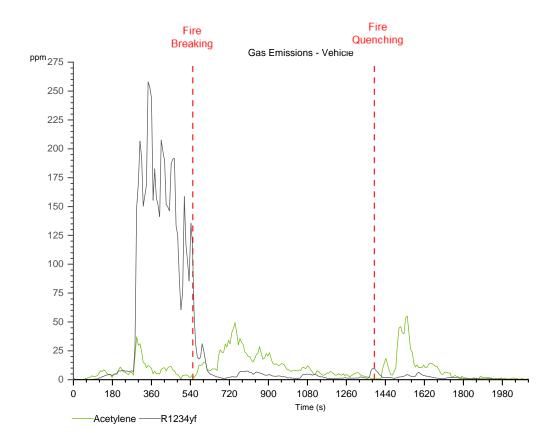


Fire breaking occurs later in Vehicle (approx. 2 minutes). Rise of HF less than 1 min after fire breaking in vehicle. Release of Acetylene and R1234yf in vehicle at around 4'30".



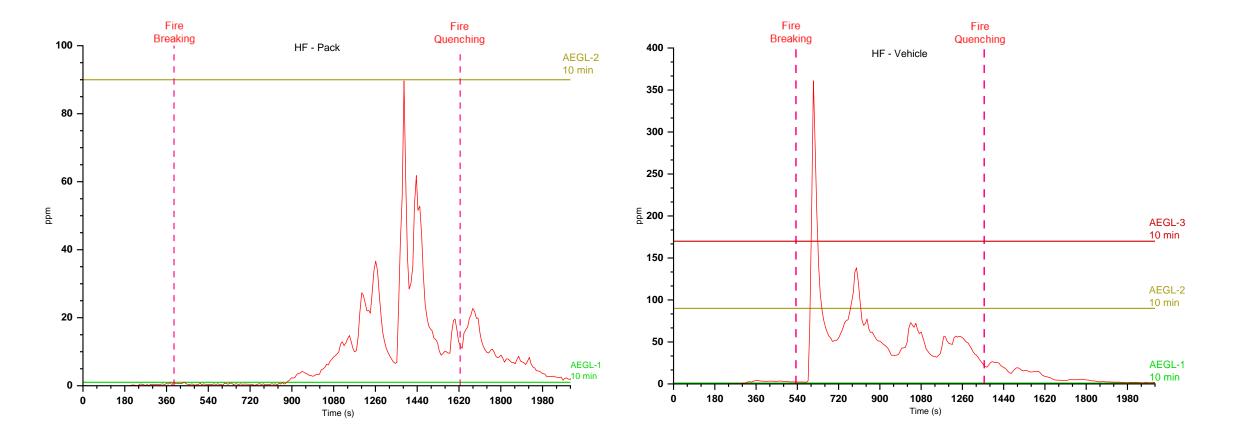
Gas Analysis – Pack vs Vehicle

- R1234yf and Acetylene are present in the emissions from the vehicle but not from pack
- R1234yf is to be expected, due to its use as a coolant gas in Air Conditioning units.
- Release of R1234yf occurs mostly during the pre-fire stage.
- Acetylene is released in bursts, with particular incidence in the combustion and fire quenching stages.



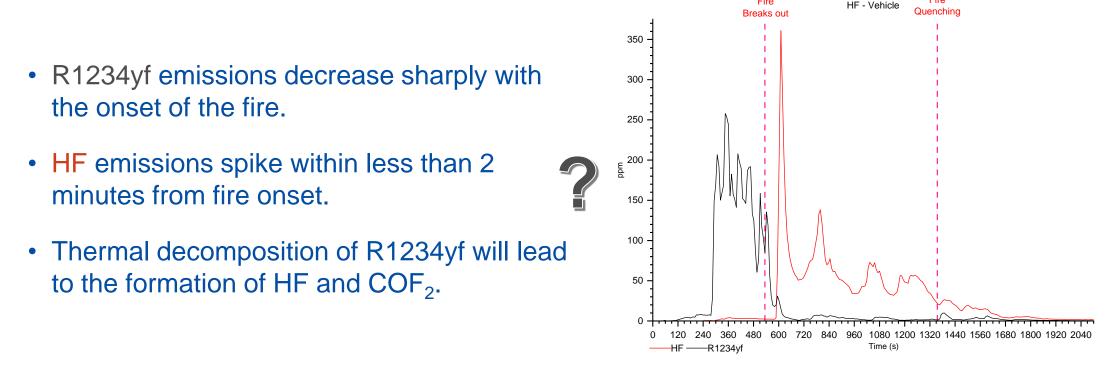


Gas Analysis – Pack vs Vehicle



Amount and dynamics of HF are different.



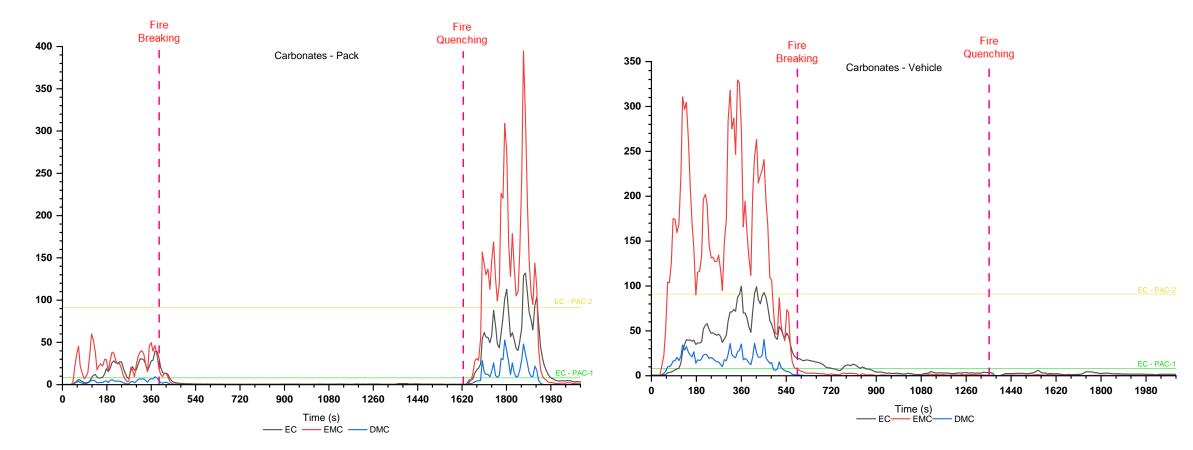


Fire

- JRC Technical and scientific support to research on safety aspects of the use of refrigerant R1234yf on MAC systems.
 <u>https://ec.europa.eu/docsroom/documents/4651/attachments/1/translations/en/renditions/native</u>
- M. Ito et al, Thermal Decomposition of Lower-GWP Refrigerants (2014), International Refrigeration and Air Conditioning Conference Paper 1538. <u>https://docs.lib.purdue.edu/cgi/viewcontent.cgi?article=2537&context=iracc</u>

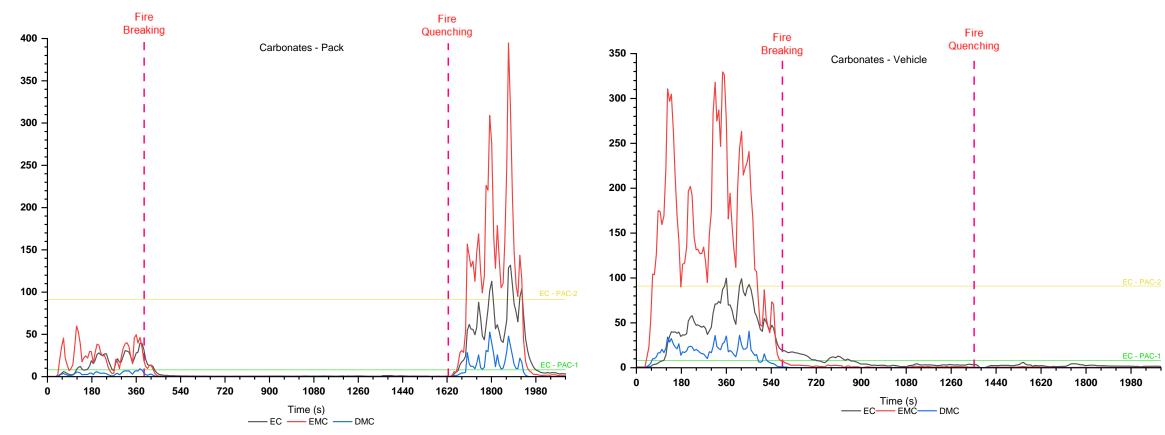
 Material Safety Datasheet for Solstice® yf Refrigerant (R-1234yf), version 6.6, revised 2023/03/20. Available in different languages and formats at <u>https://www.honeywellmsds.com/ehswww/external/search/search_index.jsp?P_LANGU=E&P_SYS=1</u>





PAC values for Ethylene Carbonate (CAS No 96-49-1) used due to being the lowest for the organic carbonates





Different emission profiles for pack and vehicle. Both with emission before fire breaking, but only pack has emission during fire quenching. Fire quenching emissions might be related to fire fighting technique.



Conclusions

- ✓ Different gas emission dynamics for pack and vehicle level in thermal propagation.
 - ✓ HF release in Vehicle happening in the early stages of fire breaking out.
 - ✓ HF release in Pack only after a few minutes of fire.
 - ✓ Much more significant release of CO in vehicle than in Pack.
 - ✓ Release of R1234yf and Acetylene in vehicle but not in Pack.
- ✓ Different composition between pack and vehicle gas emissions, starting before 5 minutes, with release of Acetylene and R1234yf.
- ✓ HF release in Vehicle above AEGL-2, and for a short time even AEGL-3, for 10 min exposure.
- ✓ Combustion of R1234yf refrigerant leads to HF formation.
- ✓ Open Path FTIR as a tool for monitoring Thermal Propagation stages, with minimal concerns for sampling.
- ? Methodological aspects are important when quantifying. (e.g. position of extraction duct)
- ? Fire quenching methods impact on composition and dynamics of gas emissions during fire fighting stage?



Acknowledgement



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Applus[⊕]





Thank you



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