

# **THERMAL PROPAGATION TOPICS**

- 1. Effect of Induced Metal Contaminants on Lithium-ion Cell Safety (Slides 2-18)*
- 2. Detection of Cell Internal Shorts (Slides 19-26)*
- 3. Industry Field Experience (Slides 27-37)*

*Presented on behalf of OICA to the EVS GTR IWG*

*March 2018*

# **EFFECT OF INDUCED METAL CONTAMINANTS ON LITHIUM-ION CELL SAFETY**

Joint investigation between GM Global Battery Systems and GM China Science Lab

Presentation to EVS GTR IWG

March 2018

Based on material originally presented at:

Advanced Automotive Battery Conference, Advanced Automotive Battery Technology, Application & Market Symposium, Session 4 – Battery Abuse-Tolerance Design and Validation, June 19, 2015, Detroit, MI

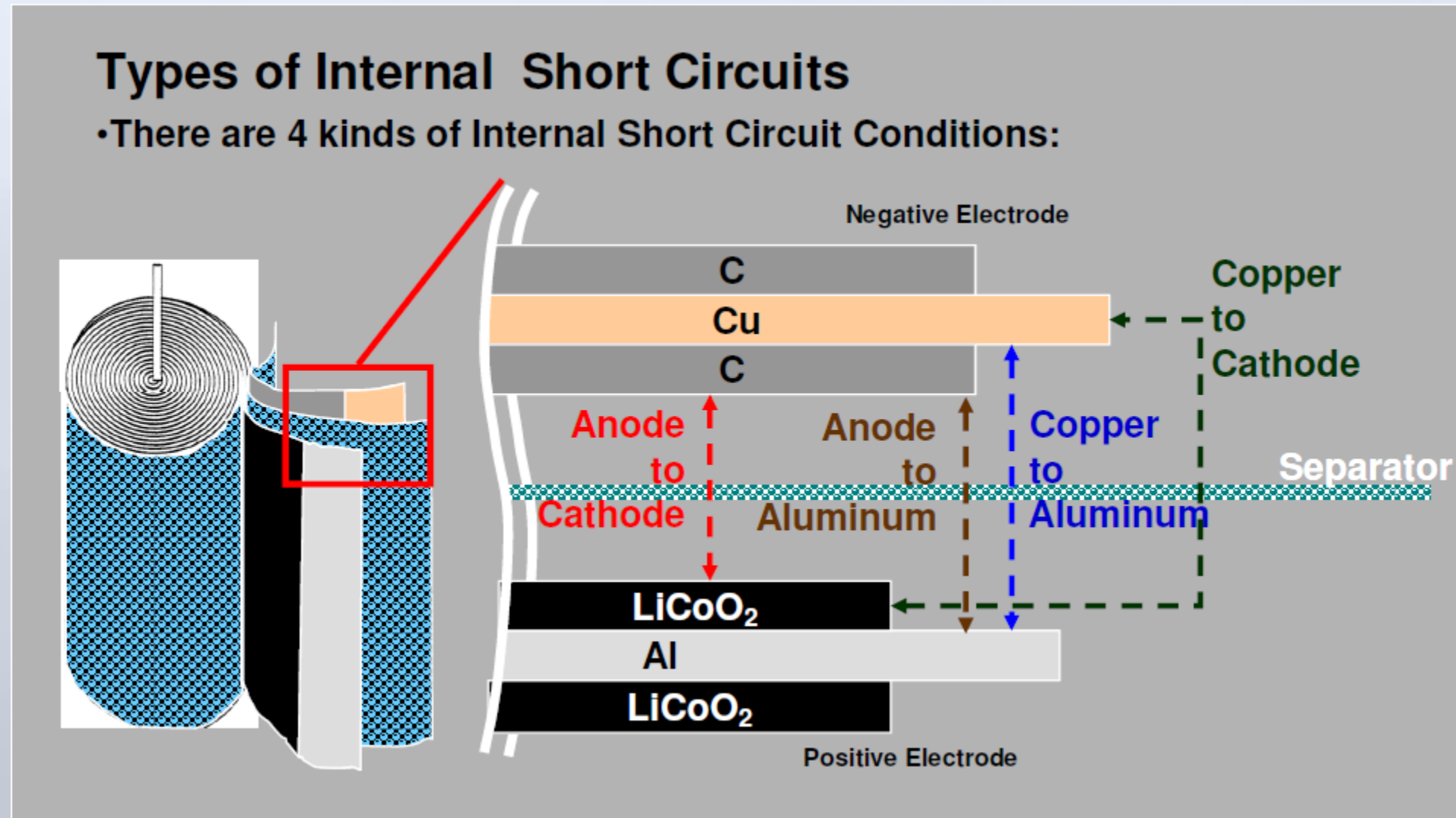
# KEY MESSAGES

- Metal particle contamination does not necessarily result in catastrophic thermal events, even after cycling under compression.
- Careful cell design (chemistry, configuration) and manufacturing process steps will minimize risk that a severe internal short circuit event can occur.

# EFFECT OF METAL PARTICLES IN LI-ION CELLS

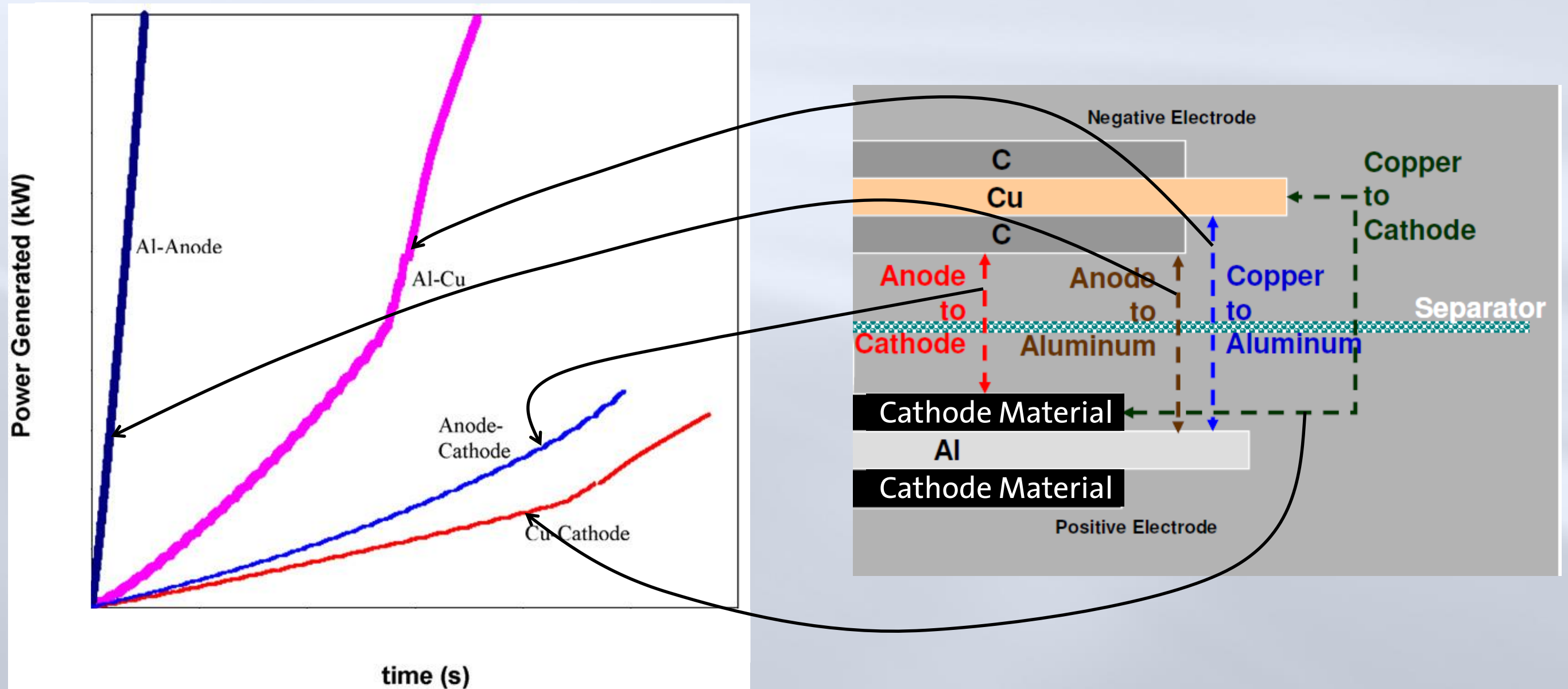
- Problem Definition – Metal particles in Li-ion cells may pose a safety risk due to the potential formation of an internal short circuit.
  - How do metal particles cause short circuits in the cells?
  - Is the severity of the short circuit dependent on the location of the particle within the cell?
    - Separator poke-through versus dendrite formation
  - Is the severity of the short circuit dependent on the size of the particle?
    - At what minimum particle size will the effects caused by the particle be detected and the cell rejected in the manufacturing process?
  - Does the severity of the short circuit increase with cycling or storage<sup>4</sup>?

# TYPES OF INTERNAL SHORTS



From "UL Transportation Tests and UL Lithium Battery Program  
Underwriter's Laboratory Inc. – General Experience and Status Update,"  
November 11, 2008

# RELATIVE RESISTANCE OF ISC TYPE



Shriram Santhanagopalan, Premanand Ramadass, John (Zhengming) Zhang,  
Analysis of internal short-circuit in a lithium ion cell,  
Journal of Power Sources, Volume 194, Issue 1, 2009, Pages 550-557.

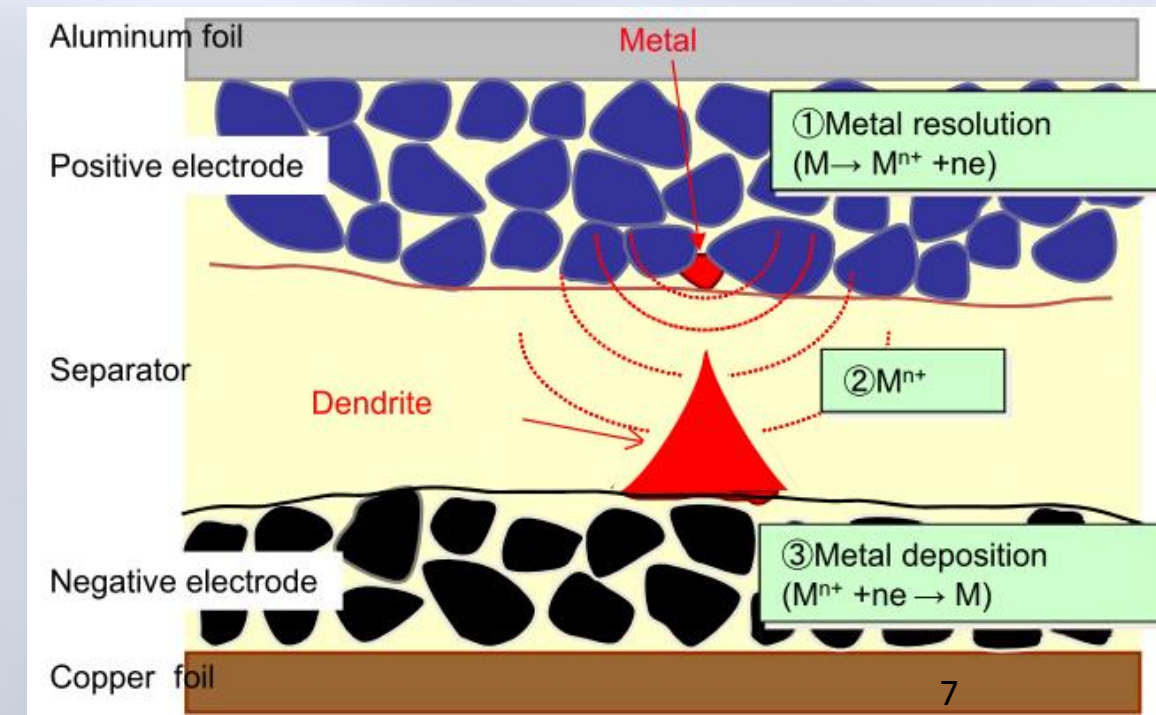
# POTENTIAL METAL PARTICLE SHORTING MECHANISMS

## ■ Separator Poke-through:

- If the particle is large enough, it could rub-through or poke-through the separator and create an internal short circuit.

## ■ Dendrite Formation:

- If the particle is located on the surface of a charged positive electrode, it could undergo electrochemical dissolution, with subsequent plating of the metal onto the surface of the negative electrode.
- If dendritic growth occurs, and the particle has enough mass, then these dendrites could grow through the separator and contact the positive electrode.

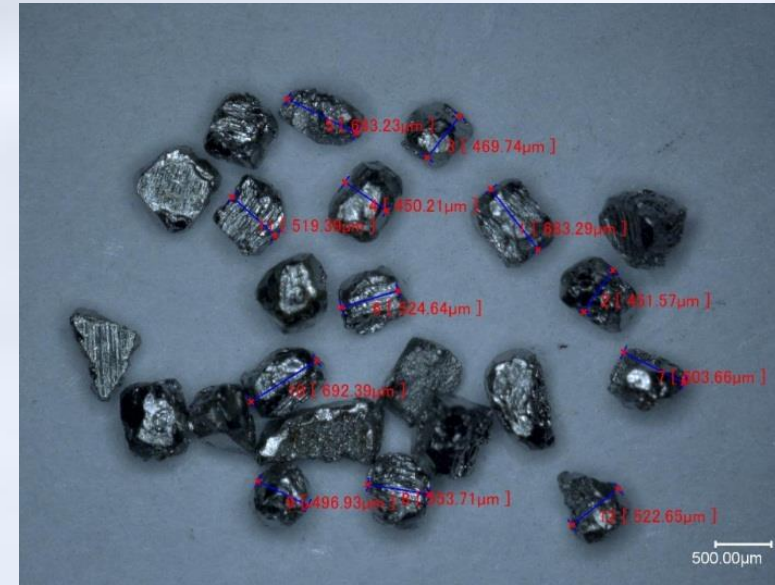




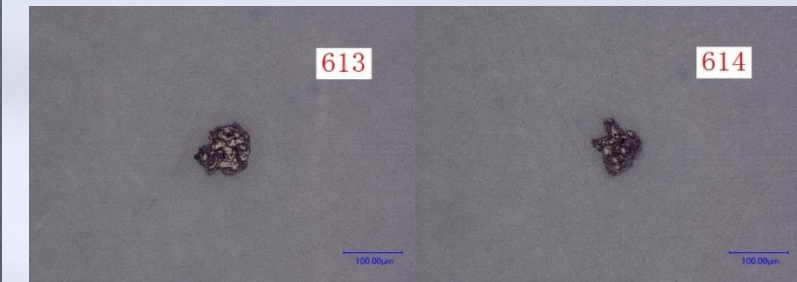
# THE STUDY:

## ■ Metal Particle Sizes

- Run 1, 500-700  $\mu\text{m}$  iron particles (ease of handling)
- Run 2, 50-150  $\mu\text{m}$  iron particles (low end of known contaminant sizes)



Run #1 Particles



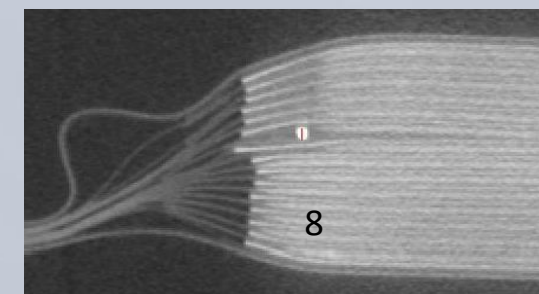
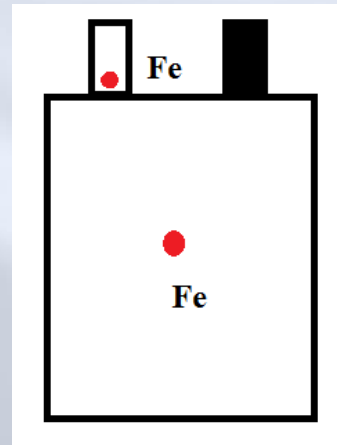
Run #2 Particles

## ■ Detection Methods and Tests

- Hi-Pot test after electrode stack assembly
- Self-Discharge (Delta-OCV) Check during Aging process step (50% SOC at 35°C for 7 days)
- Cycle Life Test (100% DOD at 1C/1C rate and 35°C under compression)
- Storage Test (100% SOC at 35°C under compression)

## ■ The Cells Used

- 1.4 Ah, multi-layer, Li-ion pouch cells (baseline and 1-particle cells, 3-6 of each type)
- MCMB graphite and NMC, with uncoated 25  $\mu\text{m}$  tri-layer separator and 1M  $\text{LiPF}_6$  EC/DEC/EMC electrolyte
- Fe particles placed in one of three locations: Anode Center, Cathode Center and Cathode Tab near the anode (see diagram and CT scan image)

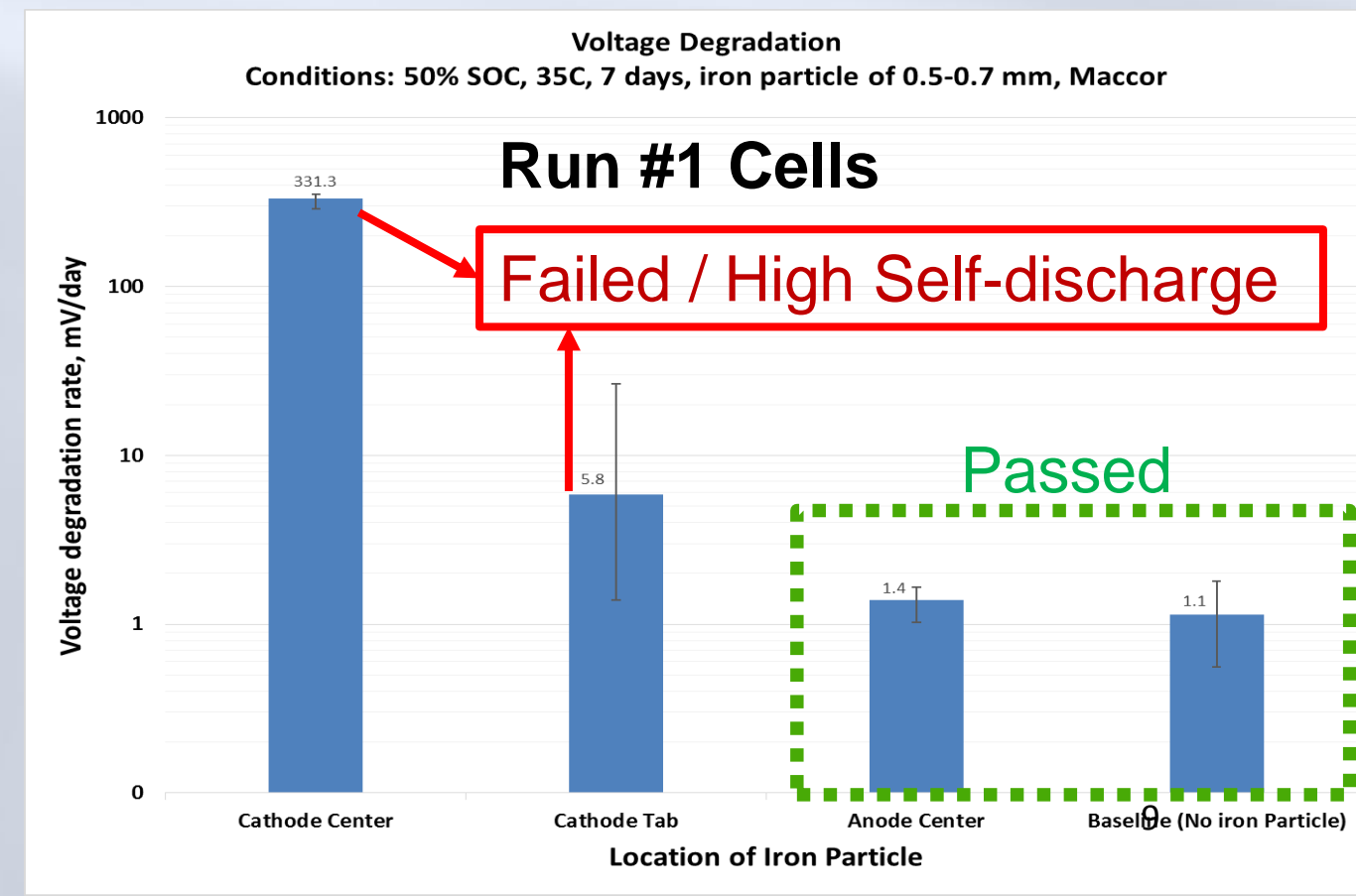




# CELL MANUFACTURING QUALITY CHECKS

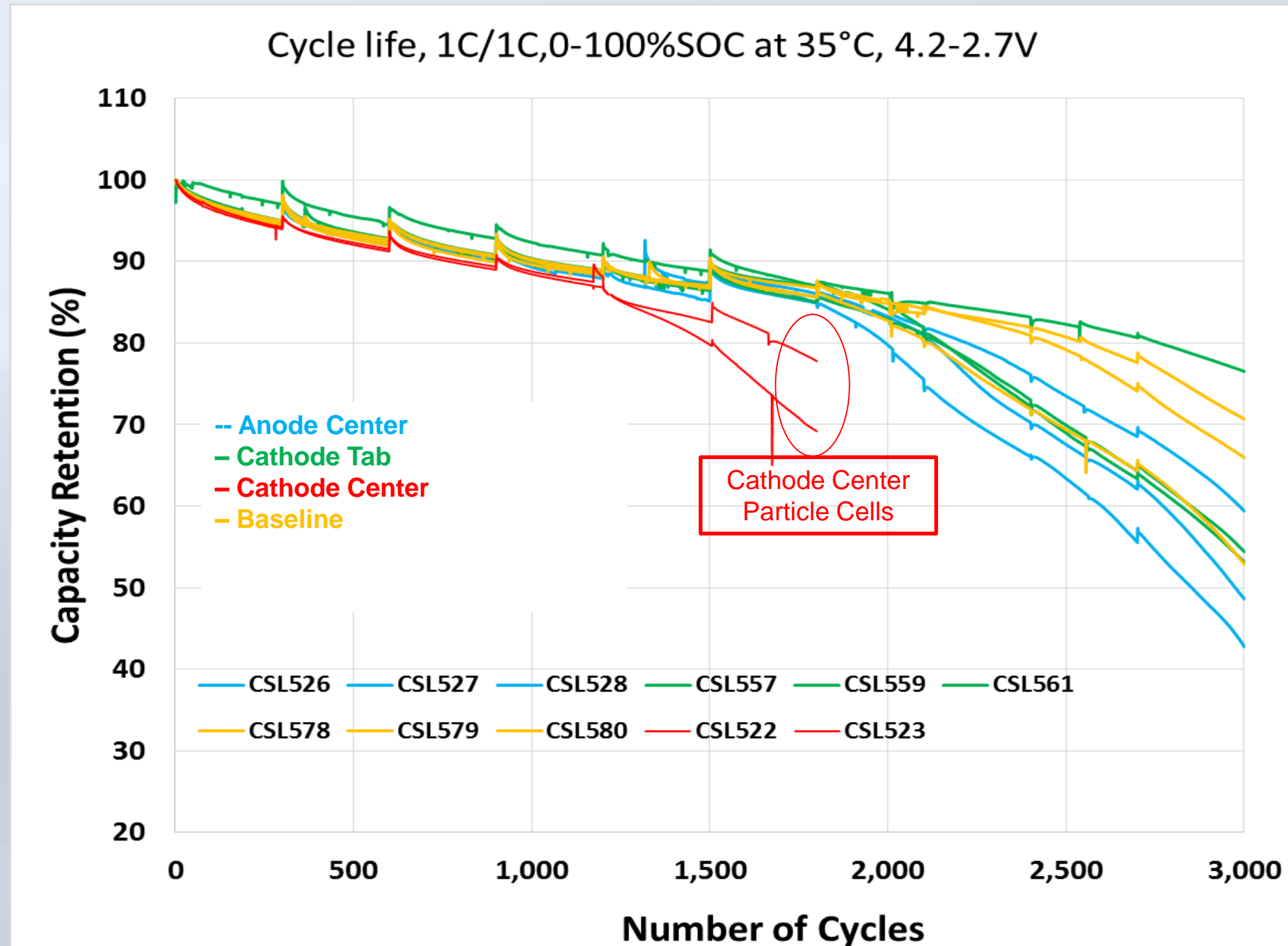
- All cells passed the Hi-Pot test after electrode stack assembly
- Only cells with large particles located on the cathode surface showed higher than normal self-discharge rates

Run	Iron Particle Size	Anode Center	Cathode Tab	Cathode Center	Baseline Cells
Run #1	> 500 $\mu\text{m}$	Pass Hi-Pot Pass self-discharge	Pass Hi-Pot Failed self-discharge	Pass Hi-Pot Failed self-discharge	Pass Hi-Pot Pass self-discharge
Run #2	100-150 $\mu\text{m}$			Pass Hi-Pot Pass self-discharge	Pass Hi-Pot Pass self-discharge
	50-100 $\mu\text{m}$			Pass Hi-Pot Pass self-discharge	Pass Hi-Pot Pass self-discharge



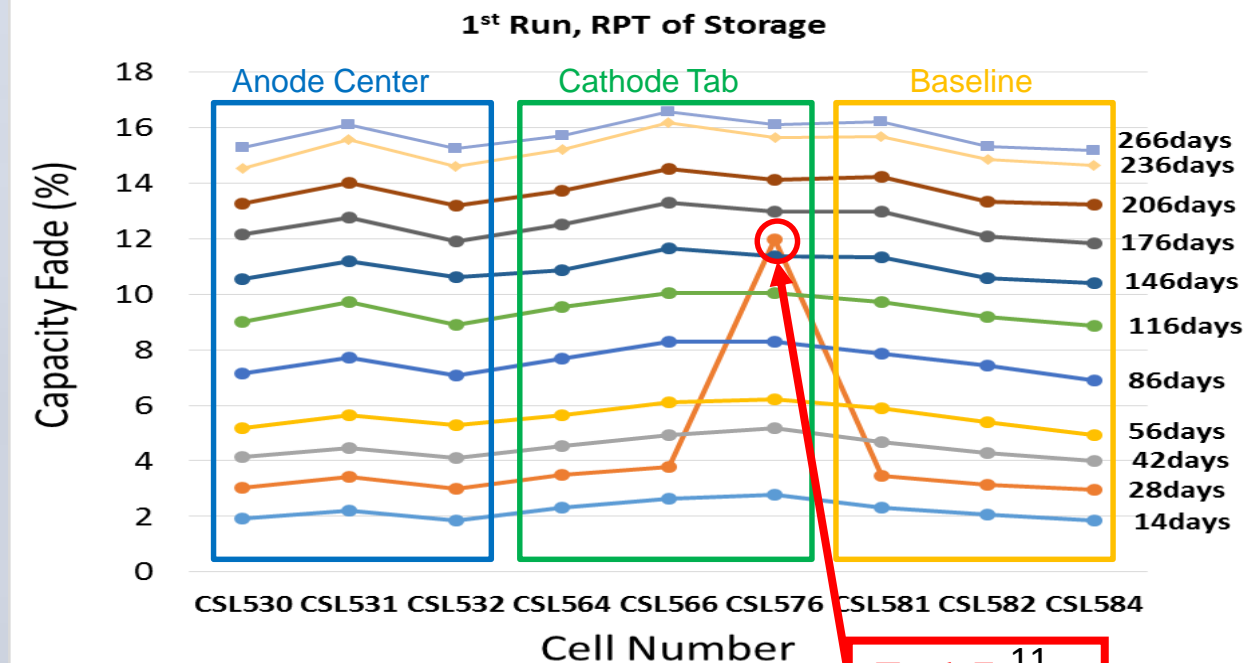
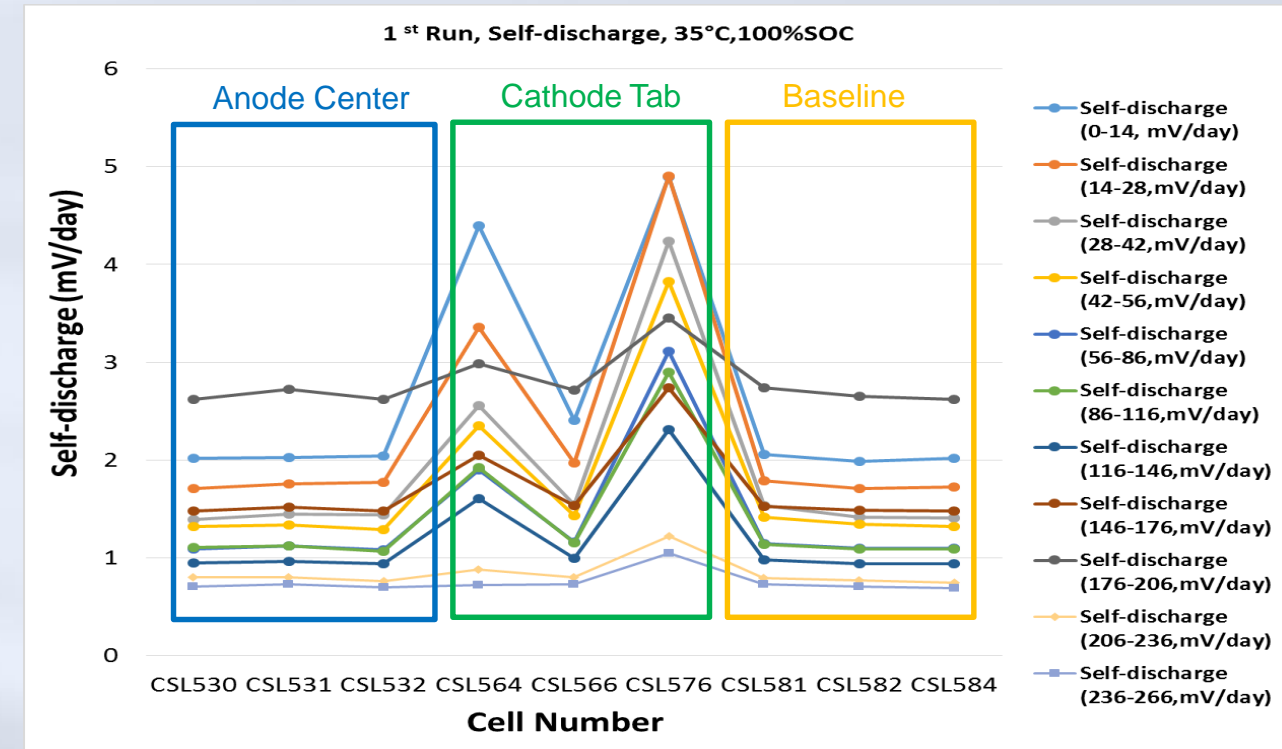
# RUN #1 CELLS – CYCLE LIFE TEST

- Cathode Center particle cells showed a more rapid Ah capacity decline.
- Anode Center and Cathode Tab particle cells showed no significant difference compared to the Baseline cells.
- No cell venting occurred.



# RUN #1 CELLS – STORAGE TEST

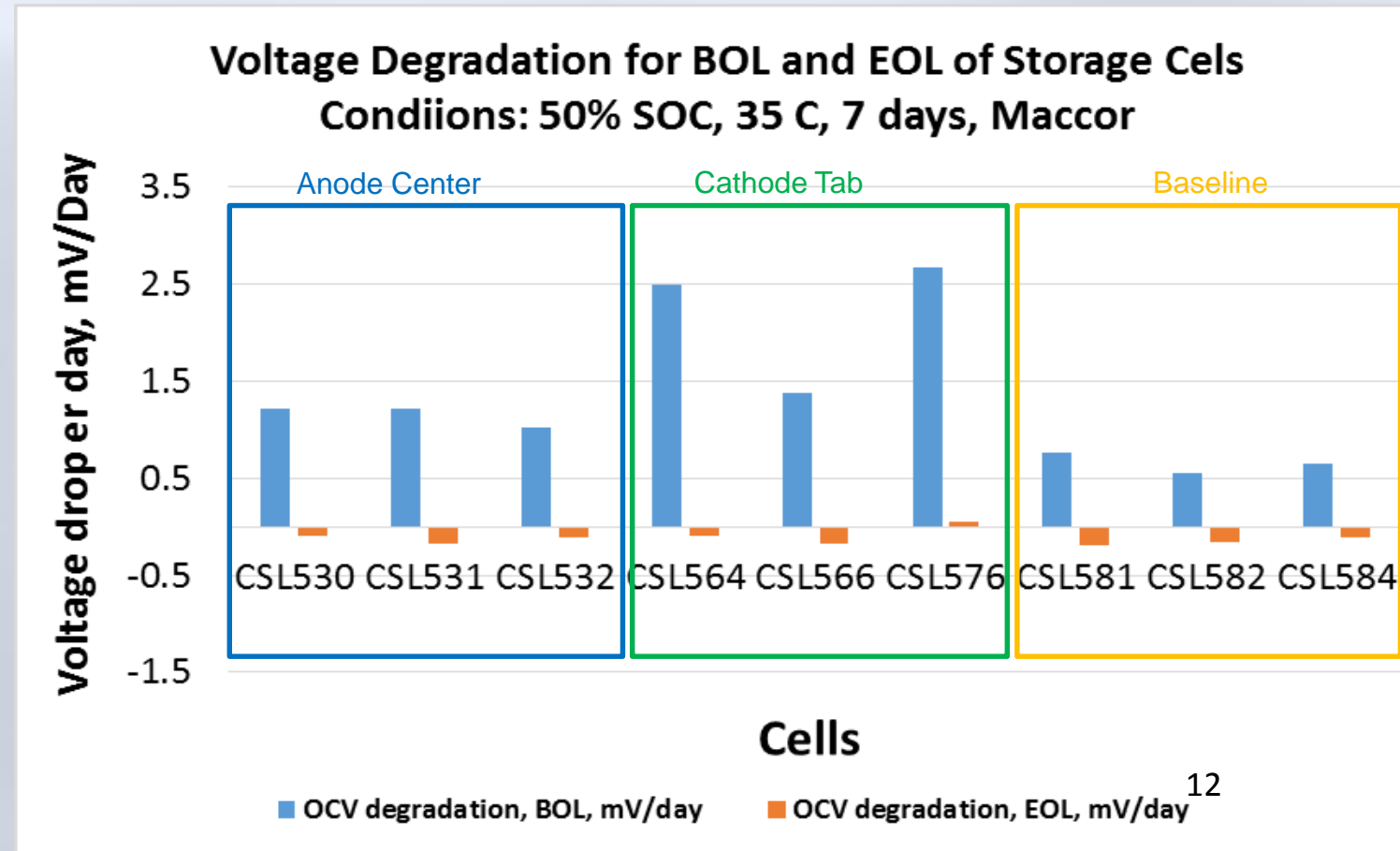
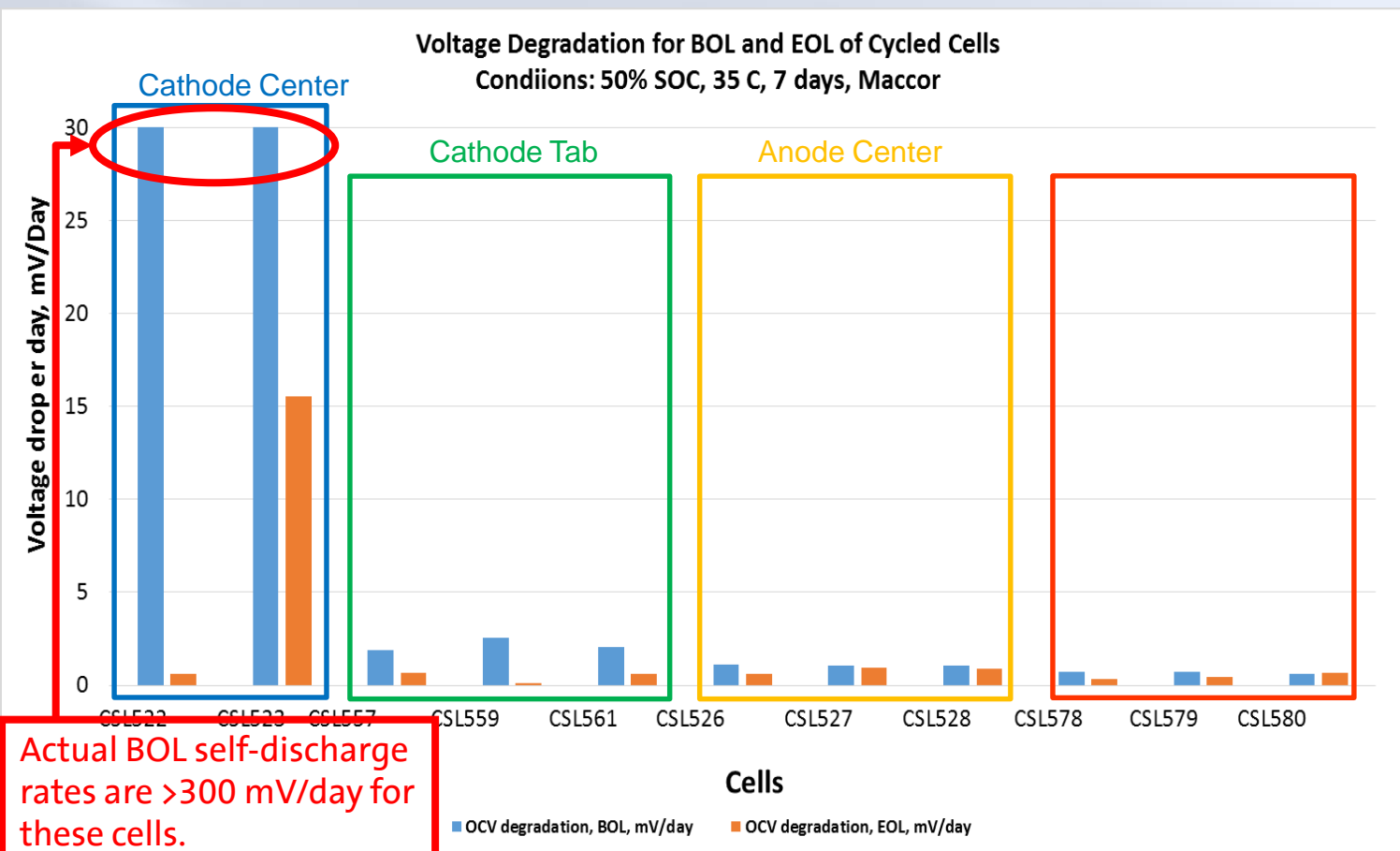
- Cathode Tab particle cells showed a higher voltage loss at each RPT.
- Anode Center particle cells showed no significant difference compared to the Baseline cells.
- Due to monthly RPT's, all cells showed similar capacity loss.
- No cell venting occurred.
- Cathode center particle cells did not undergo this test due to their already high self-discharge rates.



Test Error

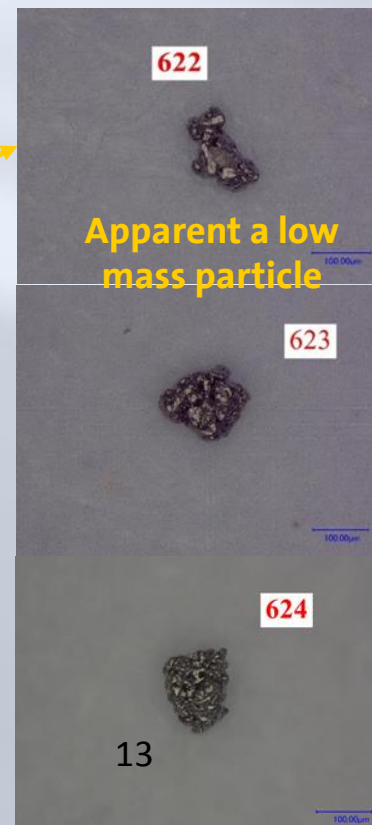
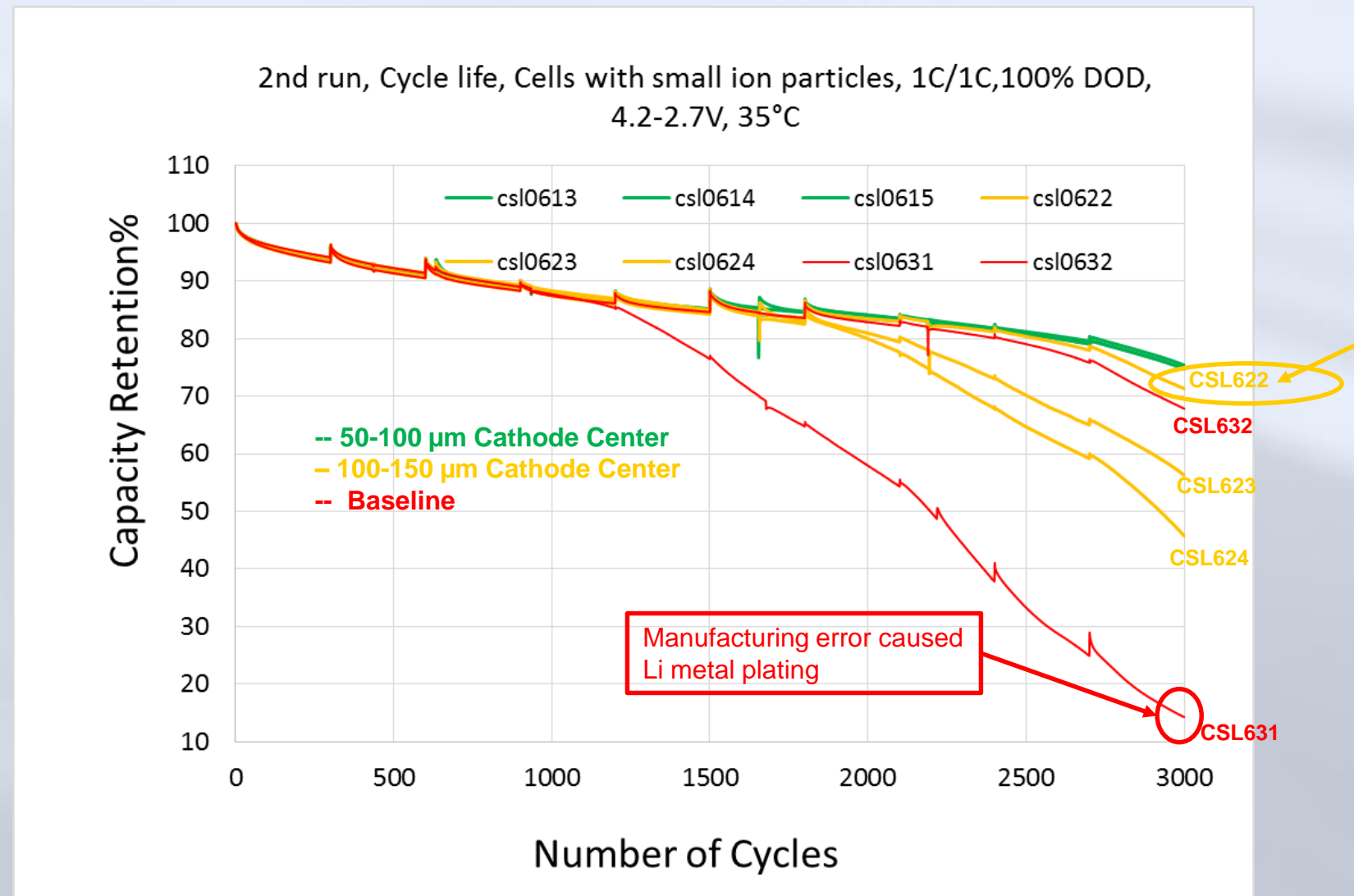
# RUN #1 CELLS – SELF-DISCHARGE RATES AFTER CYCLING OR STORAGE TESTS

- In all cases, both the particle containing cells and the baseline cells showed reduced 7-day self-discharge rates after their cycling or storage tests.



# RUN #2 CELLS – CYCLE LIFE TEST

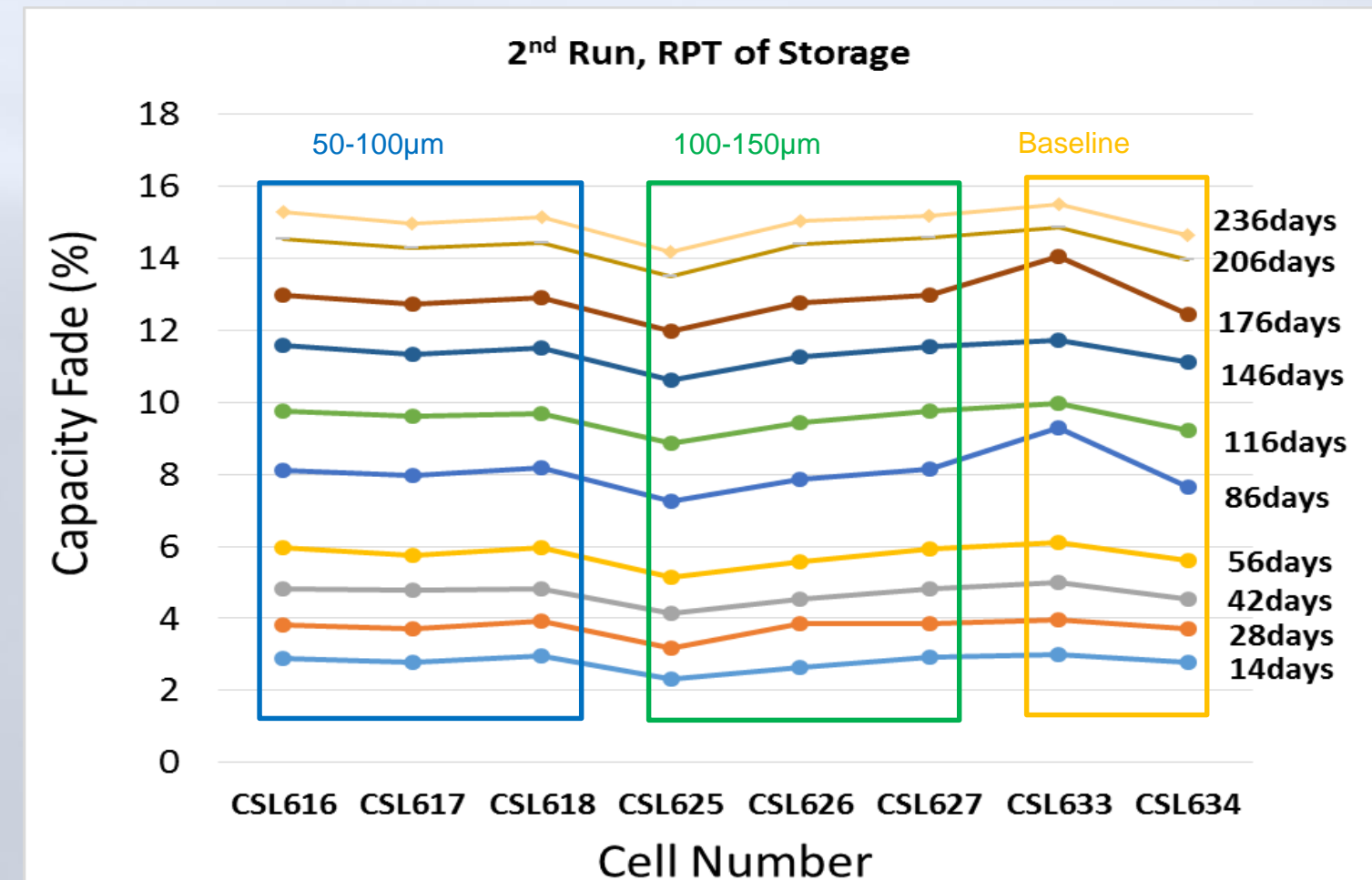
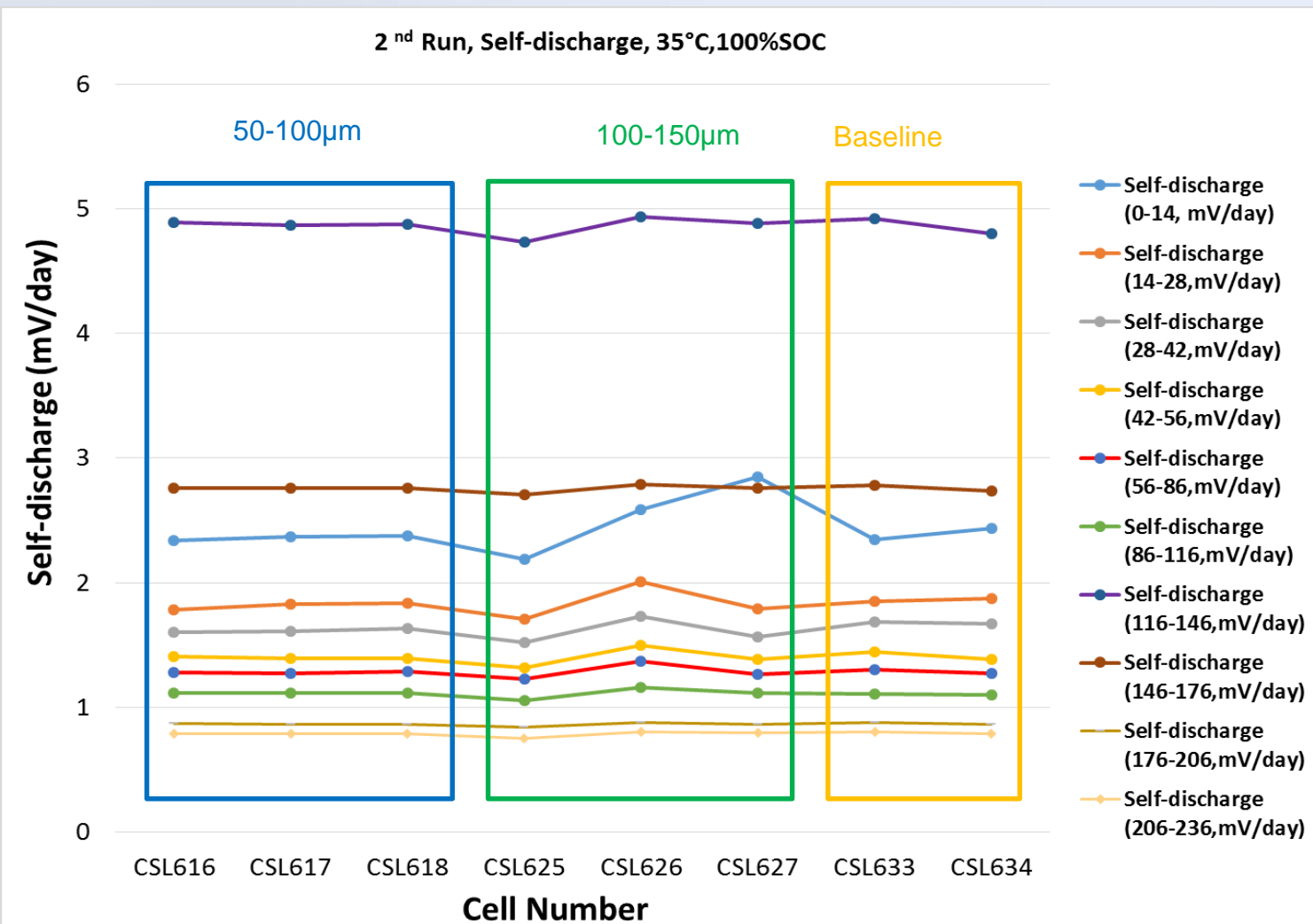
- 100-150  $\mu\text{m}$  particles appear to be the minimum size that causes premature cycle life failure.
- No cell venting occurred





# RUN #2 CELLS – STORAGE TEST

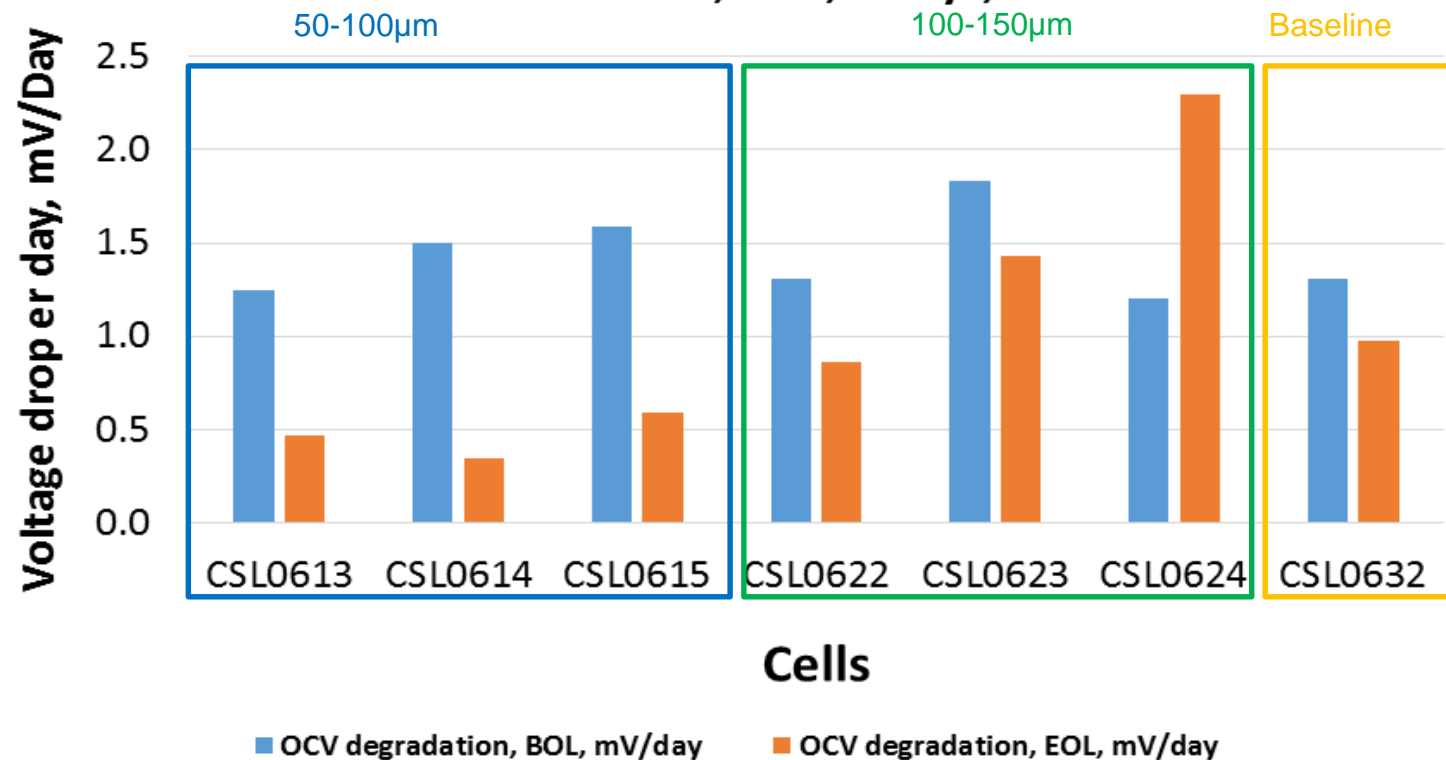
- For cells with particles <math><150 \mu\text{m}</math>, no significant self-discharge rates or capacity loss differences were seen between them and the Baseline cells.



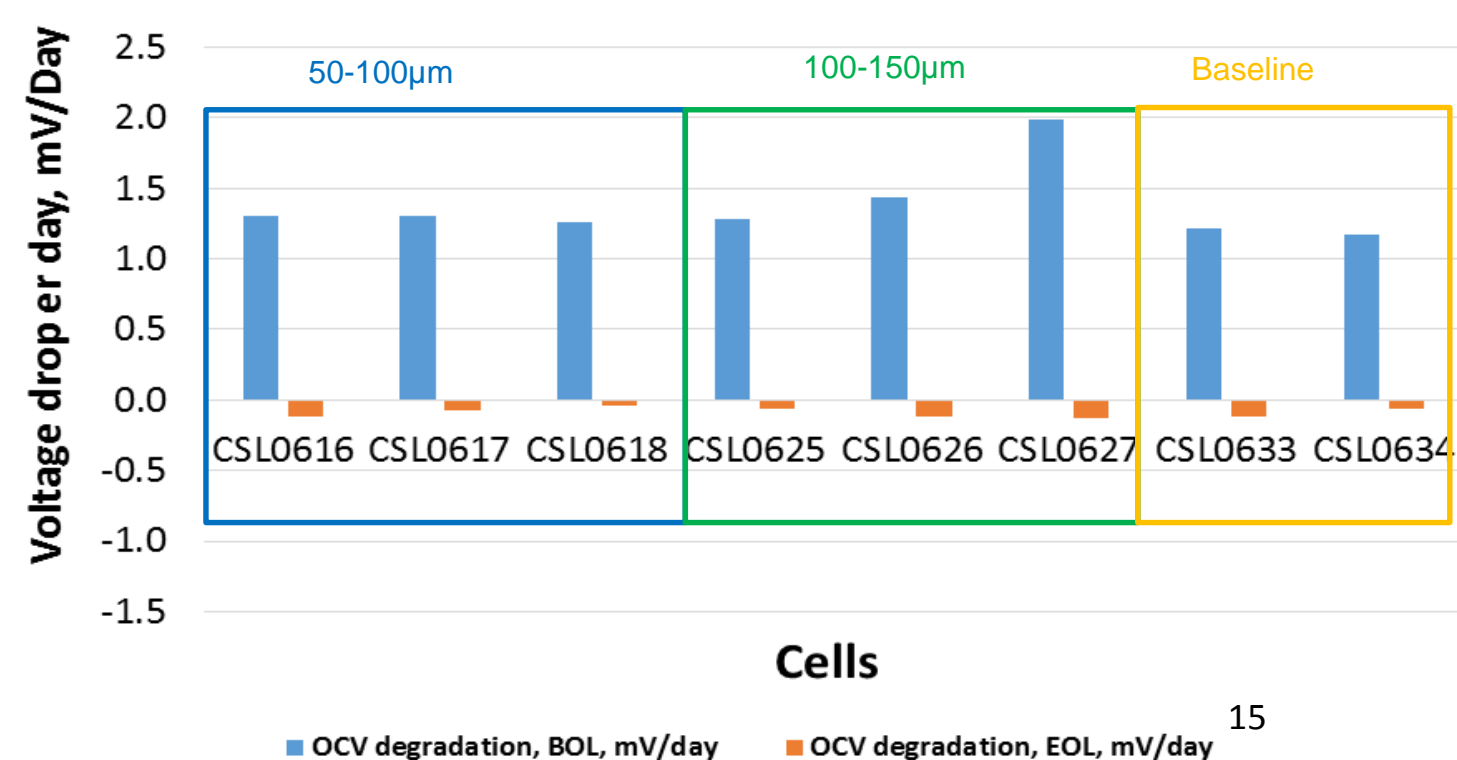
# RUN #2 CELLS – SELF-DISCHARGE RATES AFTER CYCLING OR STORAGE TESTS

- In almost all cases, both the particle containing cells and the baseline cells showed reduced 7-day self-discharge rates after their cycling or storage tests.

Voltage Degradation for BOL and EOL of Cycled Cells  
 Conditions: 50% SOC, 35 C, 7 days, Maccor

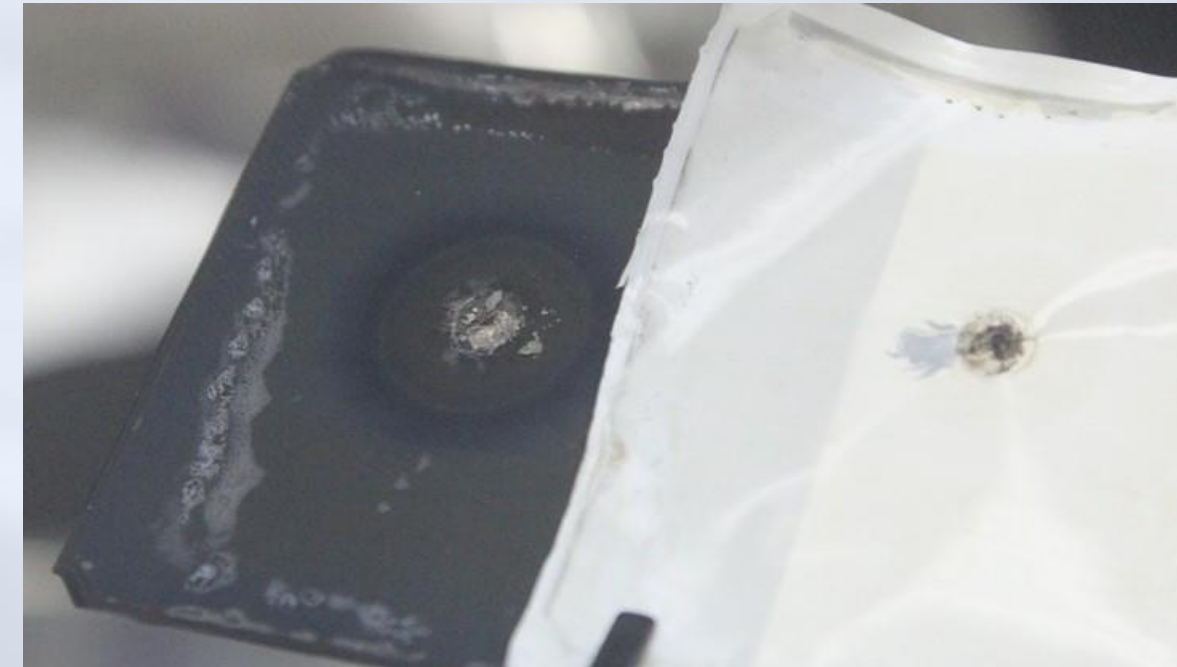


Voltage Degradation for BOL and EOL of Storage Cells  
 Conditions: 50% SOC, 35 C, 7 days, Maccor



# POST-TESTING TEARDOWN

- All particle containing and Baseline cells were examined post-test.
  - 7-day Self-Discharge test and teardown to confirm presence of the particle and root cause of failure.
  - Evidence of Fe dendrites found on separator and anode surface for all Cathode Center and Cathode Tab cells. (No particle left on cathode surface).
  - Li metal plating also found on anode surface surrounding the Fe dendrite growth (worse for large particle cells, not seen in  $<100\ \mu\text{m}$  size particle cells).
  - Original particles found in all Anode Center cells.



# TEST RESULT SUMMARY:

Run	Iron Particle Size	Particle Location	7-day Aging OCV Check	Cycling vs. Baseline Cells	Storage vs. Baseline Cells
Run #1	>500 $\mu\text{m}$	Cathode Center	Exceeded 2 mV/day Limit	More rapid capacity fade	<i>Not Run</i>
		Cathode Tab	Exceeded 2 mV/day Limit	No Difference	Greater Self-discharge
		Anode Center	Passed	No Difference	No Difference
Run #2	100-150 $\mu\text{m}$	Cathode Center	Passed	Slightly more rapid capacity fade	No Difference
	50-100 $\mu\text{m}$	Cathode Center	Passed	No Difference	No Difference

# CONCLUSIONS:

- Particles much larger than the separator's thickness (20-28 times) did not push through to create an internal short, even after cycling under compression.
- Metal particles on the negative electrode (anode) did not cause internal cell shorting.
- Only metal particles initially located on the positive electrode, if of sufficient size and mass, caused internal cell shorting.
- Large metal particles are detectable by self-discharge and capacity loss in the manufacturer's Aging/Storage process step.
- Particles below 100  $\mu\text{m}$  have limited or no effect, while particles near 150  $\mu\text{m}$  did cause pre-mature cycle life failure.



# Detection of Cell Internal Shorts

OICA Presentation to the EVS GTR IWG

March 2018

# Key Messages

- Many internal shorts can be detected during manufacture and in usage
- Internal short behavior can often be measured and understood

# Types of Shorts

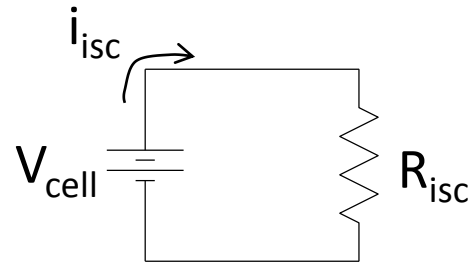
- Immediate
  - Largely managed by cell design and manufacturing control
  - Possibly detectable during manufacturing process
- Developing
  - Likely not possible to fully prevent
  - Possibly detectable during usage –
    - Requires appropriate cell measurements and diagnostic algorithms
    - Assumes that short develops over multiple cycles / days

# Possible Detection Methods

- In manufacturing process
  - Self-discharge rate – monitor for values above a defined limit
    - Times, methods, limits vary by manufacturer
  - High voltage test prior to electrolyte added
- In vehicle
  - Cell voltage – monitor for low cell voltage
  - Cell to cell voltage difference – monitor for differences above a defined limit

# Why Can These Methods Work?

- Simple circuit model



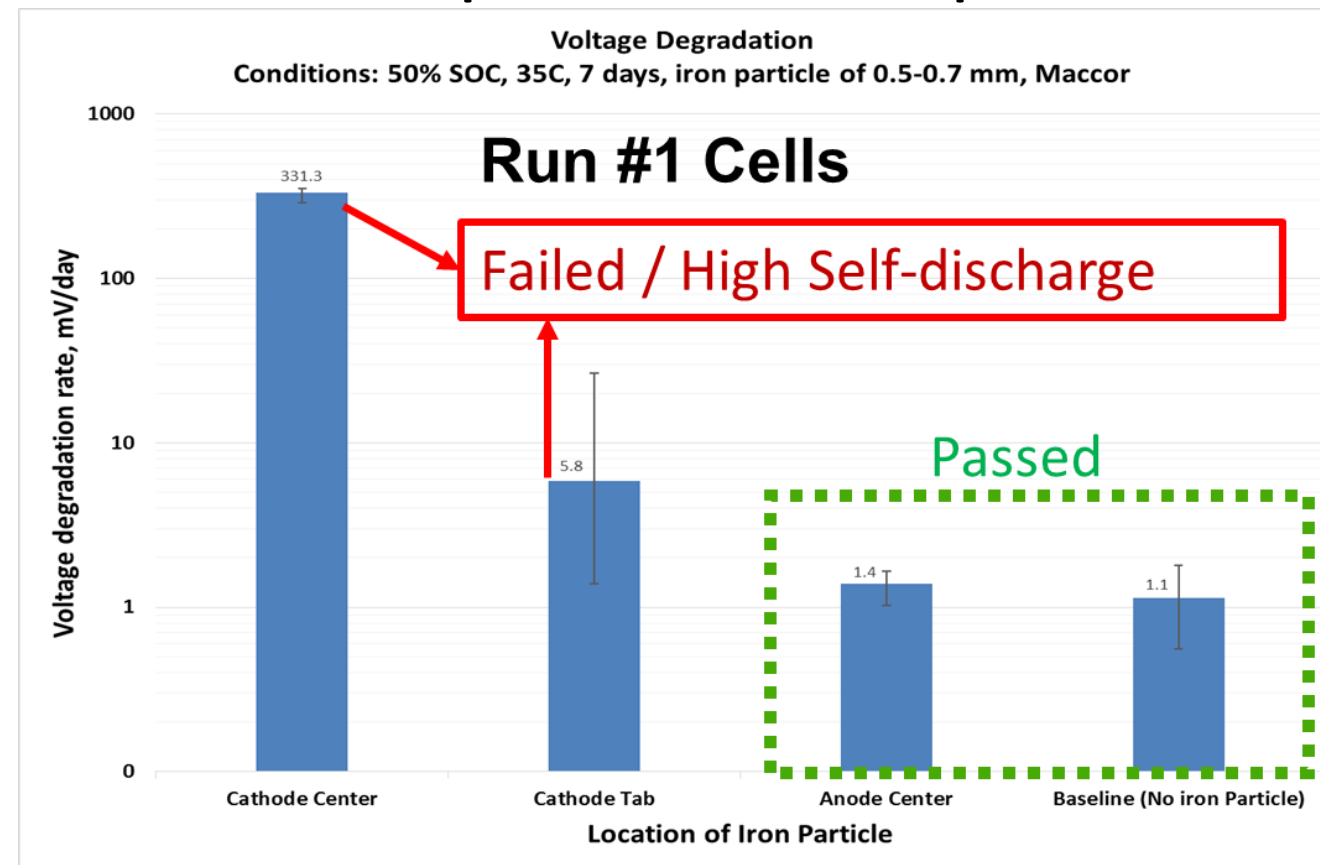
$$E_{\text{dis}} = \int V_{\text{cell}} \cdot i_{\text{isc}} \, dt$$

- Since stored energy is proportional to battery voltage, as energy is dissipated through resistance, battery voltage drops.
- Voltage reduction is observable.

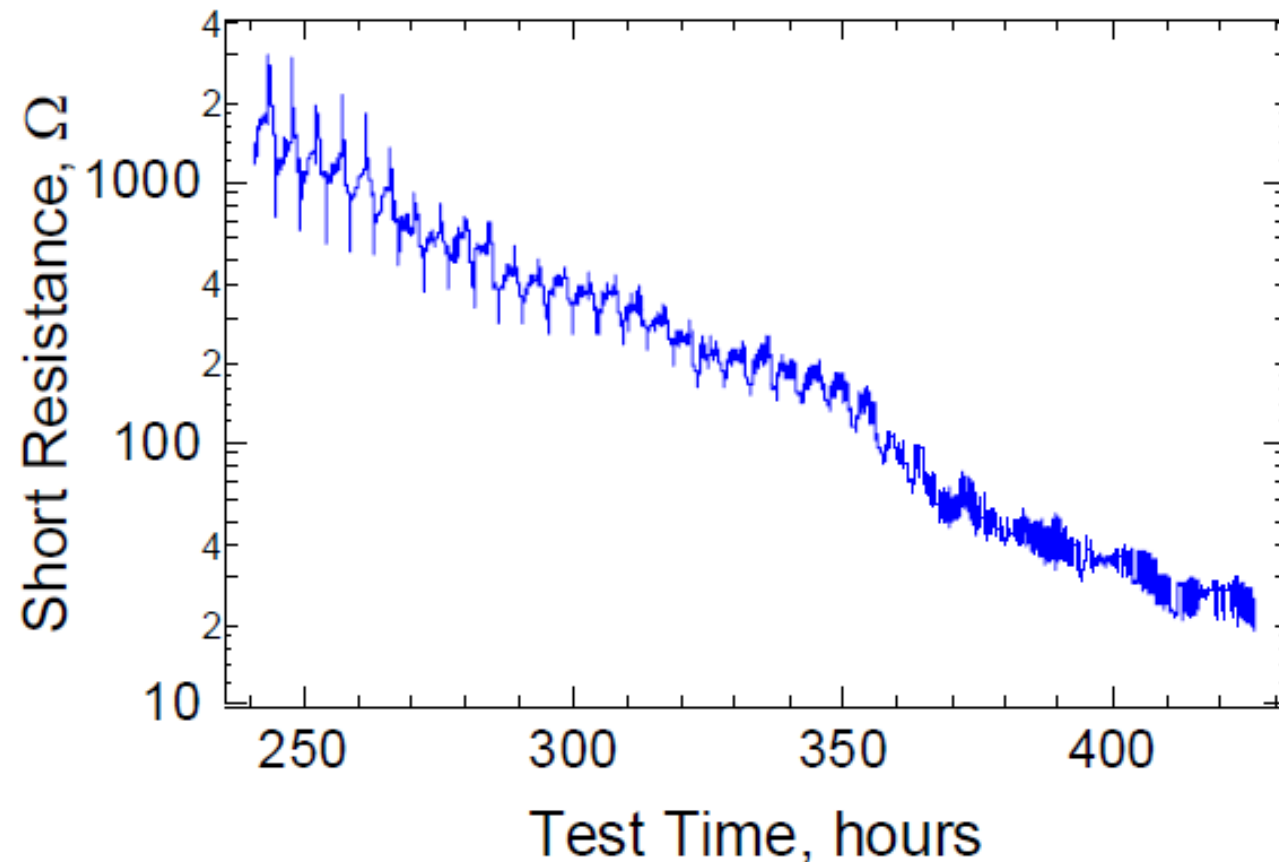


# Example of Manufacturing Process Detection

- Voltage degradation monitored for a period of time
- Measured degradation compared to acceptable limit values



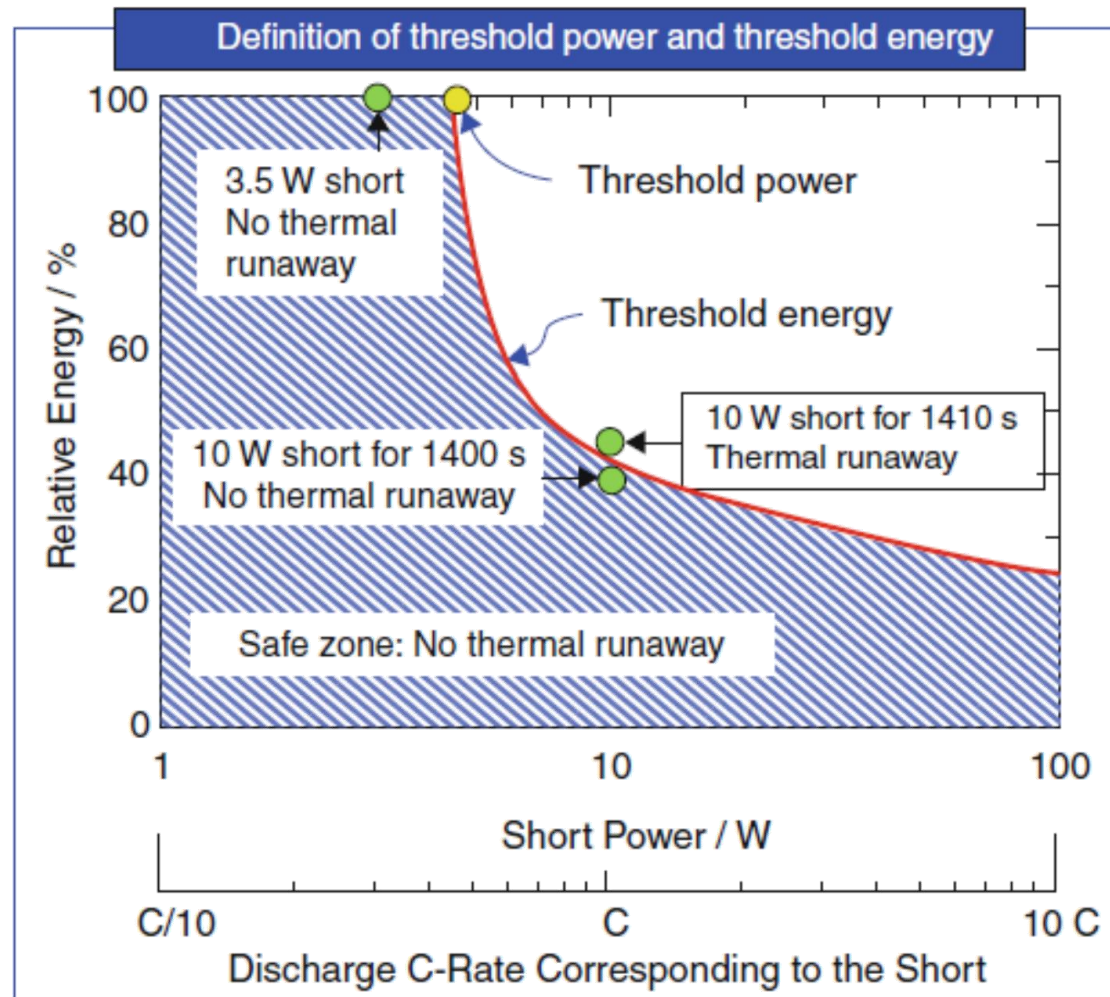
# Internal Short Resistance Behavior In Use



- Resistance reduces over time
- Cell energy ( $\therefore$  voltage) is reducing throughout time
- As energy loss integrates over time, can reach detectable level prior to short circuit at critical level

McCoy, C., Sriramulu, S., Stringfellow, R., Ofer, D. & Barnett, B. Lithium-Ion Battery Safety: Detection of Developing Internal Shorts and Suppression of Thermal Runaway. 46<sup>th</sup> Power Sources Conference, 2014.

# Critical Internal Short Resistance



- Depending on cell chemistry, cell design, and application, there exists a threshold between thermal runaway and no thermal runaway

**Fig. 9.9** The Safe Zone construct

Barnett B., Ofer D., Sriramulu S., Stringfellow R. (2013) Lithium-Ion Batteries, Safety. In: Brodd R. (eds) Batteries for Sustainability. Springer, New York, NY

# Industry Field Experience

Lithium Ion Battery Internal Short

Presentation to EVS GTR IWG

March 2018

# Key Messages

- Industry has extensive field experience with lithium ion cells
- There have been no known incidents of internal short circuits resulting in cell thermal runaway



# General Motors Experience

Lithium ion cells in field since 2010

Multiple vehicle applications

# GM battery cell varieties

- Chevrolet Volt
  - Construction type: Pouch
  - Cathode material:
    - 50% LMO / 50% 111 NMC
    - 30% LMO / 70% 442 NMC
    - 532 NMC
- Spark EV
  - Construction type: Pouch
  - Cathode material:
    - LFP (*limited volume*)
    - 30% LMO / 70% 442 NMC
- Bolt
  - Construction type: Pouch
  - Cathode material:
    - 622 NMC
- Mild Hybrid
  - Construction type: Cylindrical
  - Cathode material:
    - 111 NMC
- Strong Hybrid
  - Construction type: Prismatic can
  - Cathode material:
    - 442 NMC

# Chemistry Description

- Cathodes
  - LMO - lithium manganese oxide
    - $\text{LiMn}_2\text{O}_4$
    - Sometimes called Manganese Spinel or Li Manganese Spinel
  - NMC – lithium nickel manganese cobalt oxide
    - $\text{Li}(\text{Ni}_x\text{Mn}_y\text{Co}_z)\text{O}_2$
    - Numerical prefix refers to relative amount of each metal in the structure
  - LFP – lithium iron phosphate
    - $\text{LiFePO}_4$
- Anodes
  - Carbon
    - Mixtures of various types of carbons (hard and soft) and/or graphites (artificial and natural with an amorphous carbon coating).

# GM estimated vehicle volumes

- Volt Gen 1 (and related vehicles): ~ 90k
- Volt Gen 2 (and related vehicles): ~ 45k
- Spark EV: ~ 7k
- Bolt: ~ 25k

**Represents nearly 50 million cells**

# Known Internal Short Induced Thermal Events

- NONE

# Possible internal short frequency

- Varies somewhat by cell type and manufacturer
- Overall GM experience shows ~2-5 events per million cells for detected cell internal short in customer vehicles

# Ford Motor Company Experience

# Ford Battery Cell Varieties

- Focus Electric
  - Construction Type: Pouch cell
  - Cathode Material: 622 NMC
  - Anode Material: Carbon
- Fusion / Mondeo / Lincoln MKZ Hybrid
  - Construction Type: Aluminum Can
  - Cathode Material: 111 NMC
  - Anode Material: Carbon
- Fusion Energi
  - Construction Type: Aluminum Can
  - Cathode Material: 111 NMC
  - Anode Material: Carbon
- C-Max Hybrid
  - Construction Type: Aluminum Can
  - Cathode Material: 111 NMC
  - Anode Material: Carbon
- C-Max Energi
  - Construction Type: Aluminum Can
  - Cathode Material: 111 NMC
  - Anode Material: Carbon



Ford Motor Company EVs by Model Year	Approximate Sales Volume
2012-2017 Ford Focus Electric (BEV)	10,500
2013-2017 Ford Fusion / C-Max / Mondeo / Lincoln MKZ Hybrid (HEV)	313,000
2013-2017 Ford Fusion / C-Max Energi (PHEV)	90,000



Approximately 35 Million Cells and no documented thermal events due to internal short